

Pyridine A, a novel pyridine alkaloid with an azoxy moiety from sponge *Cribrochalina* sp.

Yuuko Kariya,^a Takaaki Kubota,^a Jane Fromont^b and Jun'ichi Kobayashi^{a,*}

^aGraduate School of Pharmaceutical Sciences, Hokkaido University, Sapporo 060-0812, Japan

^bWestern Australian Museum, Perth, WA 6000, Australia

Received 8 November 2005; revised 18 November 2005; accepted 24 November 2005

Available online 20 December 2005

Abstract—A novel cytotoxic bis-3-alkylpyridine alkaloid with an azoxy moiety, pyrinadine A (**1**), has been isolated from an Okinawan marine sponge *Cribrochalina* sp. (SS-1115), and the structure was elucidated by spectroscopic data and chemical means.

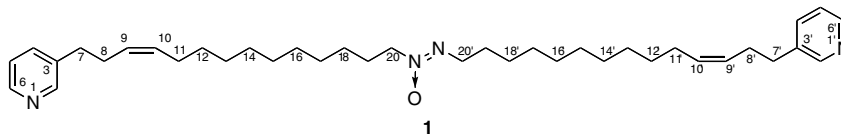
© 2005 Elsevier Ltd. All rights reserved.

Marine sponges are a rich source of bioactive secondary metabolites with unprecedented skeletons. A number of 3-alkylpyridine alkaloids have been isolated from marine sponges of several genera.¹ Most of them possess a long aliphatic chain with various nitrogen-containing terminus,² some of which have dimeric or polymeric structures of 3-alkylpyridine.³ During our continuing search for bioactive substances from marine sponges,⁴ we previously isolated cytotoxic pyridine alkaloids from sponges of the genera *Theonella*,⁵ *Nyphates*,⁶ and *Amphimedon*.⁷ Here we describe the isolation and structure elucidation of a novel cytotoxic bis-3-alkylpyridine, pyrinadine A (**1**),⁸ from the marine sponge *Cribrochalina* sp.

The sponge *Cribrochalina* sp. (SS-1115) collected off the Unten Port, Okinawa, was extracted with MeOH. EtOAc-soluble materials of the MeOH extract were subjected to a silica gel column (CHCl₃/MeOH) followed by an amino silica gel column (hexane/EtOAc) and then reversed-phase HPLC (J'sphere ODS-L80, CH₃CN/H₂O) to afford pyrinadine A (**1**, 0.00011%, wet weight).

Pyrinadine A (**1**) was revealed to have the molecular formula, C₃₈H₆₀N₄O, by HRESIMS [*m/z* 589.4818

(M+H)⁺, Δ −2.7 mmu]. The characteristic band at 1505 cm^{−1} in the IR spectrum suggested the presence of an azoxy group. Aromatic proton signals [H-2, H-2', H-6, and H-6', δ_H 8.50 (4H); H-4 and H-4', δ_H 7.72 (2H); H-5 and H-5', δ_H 7.40 (2H)] in the ¹H NMR spectrum suggested that **1** possessed two 3-alkylpyridine rings. The ¹³C NMR spectrum revealed five pairs of sp² carbon signals [C-2 and C-2', δ_C 147.0 (2C, d); C-3 and C-3', δ_C 138.2 (2C, s); C-4 and C-4', δ_C 138.7 (2C, d); C-5 and C-5', δ_C 124.1 (2C, d); C-6 and C-6', δ_C 144.6 (2C, d)] due to the two pyridine rings. The ¹H and ¹³C NMR data [H-9 and H-9', δ_H 5.33 (2H); H-10 and H-10', δ_H 5.43 (2H); C-9 and C-9', δ_C 127.0 (2C, d); C-10 and C-10', δ_C 132.0 (2C, d)] indicated the presence of two disubstituted double bonds. Thus, nine unsaturation numbers were accounted for. The ¹³C NMR spectrum showed a pair of sp³ carbon signals due to methylenes (C-20, δ_C 69.7; C-20', δ_C 52.1) at relatively lower field as compared with those of methylenes in long alkyl chains (δ_C 23–30). The chemical shifts of C-20 and C-20' indicated that these carbons were adjacent to azoxy moiety, which was the remaining part (N₂O) derived from the NMR data and molecular formula. The position of the oxygen atom in the azoxy moiety



Keywords: Sponge; Alkaloid; Pyridine; Azoxy.

* Corresponding author. Tel.: +81 11 706 4985; fax: +81 11 706 4989; e-mail: jkobay@pharm.hokudai.ac.jp

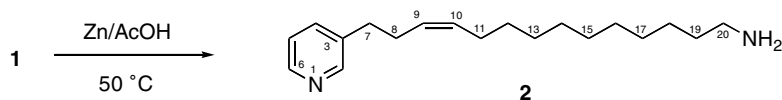


Figure 1. Reductive degradation of pyrinadine A (**1**) and the structure of its reductive product (**2**).

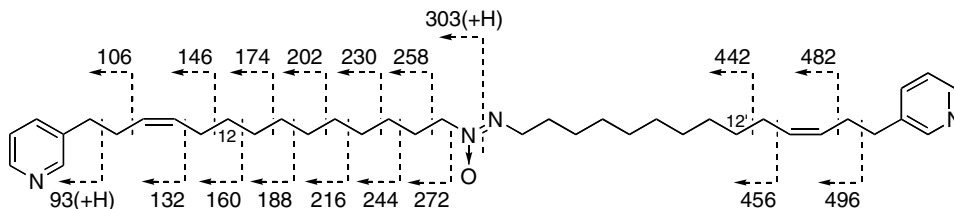


Figure 2. Fragmentation pattern of pyrinadine A (**1**) in ESI MS/MS [parent ion; m/z 589 ($M+H$)⁺].

was elucidated to be on the C-20 side on the basis of the ¹H and ¹³C NMR chemical shifts of the C-20 and C-20'.⁹ The geometry of the azoxy moiety was deduced to be *Z* from the UV absorption maximum (213 nm) of **1**, since those of the *Z*- and *E*-azoxy compounds have been observed in the range of 220 ± 3 and 230 ± 3 nm, respectively.¹⁰

The ¹H–¹H COSY, HOHAHA, and HMBC spectra revealed the connectivity from two β-substituted pyridine rings to C-12 and C-12'. *Z*-Geometry of two olefins at C-9 and C-9' was assigned from the chemical shifts of allylic carbons [C-8 and C-8', δ_C 28.4 (2C); C-11 and C-11', δ_C 27.2 (2C)].¹¹ Pyrinadine A (**1**) was treated with zinc/acetic acid (Fig. 1),¹² to give 3-alkylpyridine **2**, which was generated by cleavage at the azoxy moiety of **1**. The molecular formulae of **2** (C₁₉H₃₂N₂) was revealed from HRESIMS [**2**, m/z 289.2650 ($M+H$)⁺, Δ +0.6 mmu], and ninhydrin test indicated that **2** had an amino group. Analysis of the ESI MS/MS spectrum of **1** revealed connectivities from C-12 and C-12' to the azoxy moiety (Fig. 2). Thus, the structure of pyrinadine A was concluded to be **1**.

Pyrinadine A (**1**) is the first pyridine alkaloid with an azoxy moiety from natural origins. Pyrinadine A (**1**) showed cytotoxicity against L1210 murine leukemia (IC₅₀, 2 μg/mL) and KB human epidermoid carcinoma cells (IC₅₀, 1 μg/mL) in vitro.

Acknowledgments

We thank Z. Nagahama and K. Uehara for their help with collection of the sponge. This work was supported by Grant-in-Aid for Science Research from the Ministry of Education, Culture, Sports, Science, and Technology of Japan.

References and notes

- Blunt, J. W.; Copp, B. R.; Munro, M. H. G.; Northcote, P. T.; Prinsep, M. R. *Nat. Prod. Rep.* **2005**, *22*, 15–61.
- (a) Quiñoa, E.; Crews, P. *Tetrahedron Lett.* **1987**, *28*, 2467–2468; (b) Sakemi, S.; Totton, L. E.; Sun, H. H. *J. Nat. Prod.* **1990**, *53*, 995–999; (c) Stierle, D. B.; Faulkner, D. J. *J. Nat. Prod.* **1991**, *54*, 1134–1136; (d) Carroll, A. R.; Scheuer, P. J. *Tetrahedron* **1990**, *46*, 6637–6646; (e) Matsunaga, S.; Shinoda, K.; Fusetani, N. *Tetrahedron Lett.* **1993**, *34*, 5953–5954; (f) Wang, G.-Y.-S.; Kuramoto, M.; Uemura, D. *Tetrahedron Lett.* **1996**, *37*, 1813–1816.
- (a) Schmitz, F. J.; Hollenbeak, K. H.; Campbell, D. C. *J. Org. Chem.* **1978**, *43*, 3913–3922; (b) Fusetani, N.; Yasumuro, K.; Matsunaga, S.; Hirota, H. *Tetrahedron Lett.* **1989**, *30*, 6891–6894; (c) Talpia, R.; Rudi, A.; Ilan, M.; Kashman, Y. *Tetrahedron Lett.* **1992**, *33*, 3033–3034; (d) Davies-Coleman, M. T.; Faulkner, D. J.; Dubowchik, G. M.; Roth, G. P.; Polson, C.; Fairchild, C. *J. Org. Chem.* **1993**, *58*, 5925–5930.
- Iinuma, Y.; Kozawa, S.; Ishiyama, H.; Tsuda, M.; Fukushi, E.; Kawabata, J.; Fromont, J.; Kobayashi, J. *J. Nat. Prod.* **2005**, *68*, 1109–1110.
- Kobayashi, J.; Murayama, T.; Ohizumi, Y.; Sasaki, T.; Ohta, T.; Nozoe, S. *Tetrahedron* **1989**, *30*, 4833–4836.
- (a) Kobayashi, J.; Murayama, T.; Kosuge, S.; Kanda, F.; Ishibashi, M.; Kobayashi, H.; Ohizumi, Y.; Ohta, T.; Nozoe, S.; Sasaki, T. *J. Chem. Soc., Perkin Trans. 1* **1990**, 3301–3303; (b) Kobayashi, J.; Zeng, C.-M.; Ishibashi, M.; Shigemori, H.; Sasaki, T.; Mikami, Y. *J. Chem. Soc., Perkin Trans. 1* **1992**, 1291–1294.
- (a) Tsuda, M.; Hirano, K.; Kubota, T.; Kobayashi, J. *Tetrahedron Lett.* **1999**, *40*, 4819–4820; (b) Hirano, K.; Kubota, T.; Tsuda, M.; Mikami, Y.; Kobayashi, J. *Chem. Pharm. Bull.* **2000**, *48*, 974–977; (c) Ishiyama, H.; Tsuda, M.; Endo, T.; Kobayashi, J. *Molecules* **2005**, *10*, 312–316.
- Pyrinadine A* (**1**). A colorless oil; UV (MeOH) λ_{max} 213 (ε 10,600), 251 (3600), 257 (4200), 262 (4700), and 269 (3700) nm; IR (KBr) ν_{max} 2924, 2853, and 1505 cm⁻¹; ¹H (CDCl₃) δ 8.50 (4H, m), 7.72 (2H, m), 7.40 (2H, m), 5.43 (m), 5.33 (m), 4.14 (2H, t, *J* = 7.3 Hz), 3.40 (2H, t, *J* = 7.2 Hz), 2.73 (4H, t, *J* = 7.6 Hz), 2.38 (4H, dt, *J* = 7.6 and 7.2 Hz), 1.95 (2H, m), 1.95 (2H, m), 1.90 (4H, m), 1.69 (2H, m), 1.0–1.4 (28H, m); ¹³C NMR (CDCl₃) δ 147.0 (2C, d), 144.6 (2C, d), 138.7 (2C, s), 132.0 (2C, d), 127.0 (2C, d), 124.1 (2C, d), 69.7 (t), 52.1 (t), 33.0 (2C, t), 28.4 (2C, t), 27.2 (2C, t), 26–30 (16C, t); ESIMS (pos.) m/z 589 ($M+H$)⁺; HRESIMS m/z 589.4818 ($M+H$)⁺, Δ –2.7 mmu.
- Yamazaki, S. *Bull. Chem. Soc. Jpn.* **1997**, *70*, 877–883.
- Swigert, J.; Taylor, K. G. *J. Am. Chem. Soc.* **1971**, *93*, 7337–7338.
- Vysotskii, M. V.; Imbs, A. B.; Popkov, A. A.; Latyshev, N. A.; Svetachev, V. I. *Tetrahedron Lett.* **1990**, *31*, 4367–4370.
- (a) Chavez, D.; Hill, L.; Hiskey, M.; Kinkead, S. *J. Energy Mater.* **2000**, *18*, 219–236; (b) Bhushan, R. G.; Vince, R. *Bioorg. Med. Chem.* **2002**, *10*, 2325–2333.